## Some Enantioselective Photocyclization Reactions in Inclusion **Crystals with Optically Active Host Compounds**

Fumio Toda,\* Hisakazu Miyamoto, Koichi Takeda, Rikiya Matsugawa, and Nobuyuki Maruyama

Department of Applied Chemistry, Faculty of Engineering, Ehime University, Matsuyama, Ehime 790, Japan

Received April 19, 1993<sup>®</sup>

Photoirradiation of inclusion crystals of 2-[N-(2-propenyl)amino]cyclohex-2-enones, 3-oxo-2cyclohexenecarboxamides, and 4-(3-butenyl)cyclohexa-2,5-dien-1-ones with the optically active host compound (R,R)-(-)-trans-4,5-bis(hydroxydiphenylmethyl)-1,4-dioxaspiro[4.4]nonane or (R,R)-(-)-trans-2,3-bis(hydroxydiphenylmethyl)-1,4-dioxaspiro[5.4]decane in the solid state gave optically active 9-azatricyclo[5.2.1.0<sup>1,6</sup>]decan-2-ones, 2-aza-1,5-dioxospiro[3.5]nonanes, and 1-carbomethoxytricyclo[4.3.1.07,10]dec-2-en-4-ones, respectively.

It has been reported that the photoreaction of the 2-[N-(2-propenyl)amino]cyclohex-2-enone derivatives 1, 3-oxo-2-cyclohexenecarboxamides 7, and 4-carbomethoxy-4-(3butenyl)cyclohexa-2,5-dien-1-ones 12 in solution gives the racemic photocyclization products 9-azatricyclo [5.2.1.0<sup>1,6</sup>]decan-2-ones 2,1-32-aza-1,5-dioxaspiro[3.5] nonanes 9,4 and 1-carbomethoxytricyclo[4.3.1.07,10]dec-2-en-4-ones 13,5 respectively. In order to synthesize the optically active photocyclization products (2, 9, and 13) selectively, we studied solid-state photoreactions of 1, 7, and 12 in their inclusion crystals with the optically active host compound such as (R,R)-(-)-trans-4,5-bis(hydroxydiphenylmethyl)-1,4-dioxaspiro[4.4] nonane (3a) or (R,R)-(-)-trans-2,3-bis-(hydroxydiphenylmethyl)-1,4-dioxaspiro[5.4]decane (3b).<sup>6,7</sup>

Recrystallization of 3a or 3b and 2-[N-benzoyl-N-(2propenyl)amino]cyclohex-2-enone (1a) from ether or toluene-hexane gave a 2:1 inclusion crystal of 3a or 3b with 1a. Photoirradiation of a suspension of a powdered 2:1 inclusion crystal of 3a with 1a in water containing sodium alkylsulfate as a surfactant for 17 h under stirring gave (+)-2a in 64% yield, after purification by column chromatography on silica gel. However, an accurate  $[\alpha]_D$  value of the (+)-2a was not determined because (+)-2a forms hydrate 4a and hemiacetal 5a in the presence of water and alcohol, respectively, and then (+)-2a shows mutarotations in the measurement of  $[\alpha]_D$  value. Easy formation of a hydrate of rac-2a has been reported.<sup>3</sup> Since formation of the hydrate of 2 is relatively slower in dry CCl<sub>4</sub>, the  $[\alpha]_D$ values of 2 measured in that solvent are shown in Table I. Furthermore, optical purity of (+)-2a is not determined by <sup>1</sup>H NMR spectral measurement in the presence of a chiral shift reagent and by HPLC analysis on a chiral solid phase because the mutarotation occurs by the formation of the hydrate of hemiacetal during the measurement or analysis. Finally, the optical purity of the (+)-2a was determined to be 99% ee for its ethylene glycol acetal, 2,2-(ethylenedioxy)-9-benzoyl-9-azatricyclo[5.3.1.0<sup>1,6</sup>]-

• Abstract published in Advance ACS Abstracts, October 1, 1993.

Anould, J. C.; Cossy, J.; Pete, J. P. Tetrahedron 1980, 36, 1585.
 Ikeda, M.; Takahashi, M.; Ohno, K.; Tamura, Y.; Kido, M. Chem.

Pharm. Bull. 1982, 30, 2269.

(3) Ikeda, M.; Uchino, T.; Takahashi, M.; Ishibashi, H.; Tamura, Y.; Kido, M. Chem. Pharm. Bull. 1985, 33, 3279.

(4) Blanc, S. L.; Pete, J.-P.; Piva, O. Tetrahedron Lett. 1992, 33, 1993.
 (5) Schultz, A. G.; Plummer, M.; Taveras, A. G.; Kulling, R. K. J. Am.

- Chem. Soc. 1988, 110, 5547. (6) Seebach, D.; Beck, A. K.; Imwinkelried, R.; Roggo, S.; Wonnacott, (7) Toda, F.; Tanaka, K. Tetrahedron Lett. 1988, 29, 551.

Table I. Photocyclization of 1 in a 2:1 Inclusion Compound with 3 in a Water Suspension

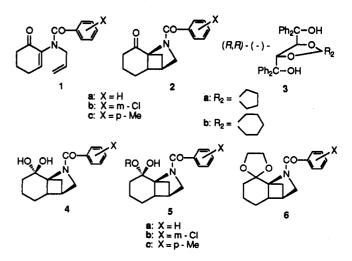
				photocyclizn product					
	st-guest	mp (°C)	irradn time (h)	product	yield (%)	mp (°C)	$ \begin{matrix} [\alpha]_{\rm D}  ({\rm deg}) \\ (c  {\rm in}  {\rm CCl}_4)^b \end{matrix} $		
3a 3b 3b 3b	1a 1a 1b 1c	118-119 139.5-142 nc <sup>a</sup> nc <sup>a</sup>	17 15 36 43	(+)-2a (+)-2a (+)-2b (+)-2c	64 81 71 65	106-107 106-107 116.5-118 162-163	+70 (0.10) +90 (1.32) +77 (0.85) +61 (0.53)		

<sup>a</sup> Not clear. <sup>b</sup> Since 2 shows mutarotation in the  $[\alpha]_D$  measurement in protic and wet solvent by formation of acetal, hemiacetal, and hydrate, the  $[\alpha]_D$  value was measured in dry CCl<sub>4</sub> in which the mutarotation occurs relatively slower.

Table II. Yield, Melting Point, and  $[\alpha]_D$  Value of (-)-6

6	yield (%)	mp (°C)	[α] <sub>D</sub> (deg) (c in CCl <sub>4</sub> )	optical purity (% ee)	
6a	72	108-110	-166 (2.53)	99	
6b	76	143-145	-144 (1.38)	>99.9	
6c	68	115-116	-151 (1.43)	99	

decane ((-)-6a), by HPLC analysis on the chiral solid phase, Chiralcel OJ (Table II).



Photoirradiation of a 2:1 inclusion crystal of 1a with 3b by the same procedure described above gave (+)-2a in 81% yield, from which (-)-6a of 99% ee was derived in 45% yield (Tables I and II). Since the host 3b was more effective than 3a for the enantioselective photocyclization of 1a, the photocyclization of 1b and 1c was carried out by using the host 3b, and (+)-2b and (+)-2c were obtained

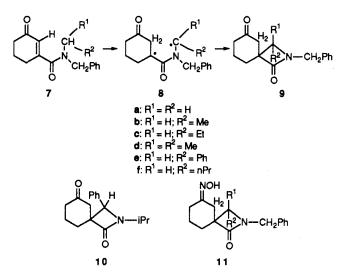
© 1993 American Chemical Society

in 71 and 65% yields, respectively (Table I). The optical purity of (+)-2b and (+)-2c was determined to be >99.9 and 99% ee, respectively, by HPLC analysis of their acetal derivatives, (-)-6b and (-)-6c (Table II).

There is a possibility that the optical purity of the acetal (-)-6 was raised up to 99->99.9% ee by an optical resolution during the course of acetalization of the (+)-2 and purification of the resulted (-)-6. Therefore, enantioselectivity of the photocyclization of 1 in the inclusion crystal with 3 cannot simply be determined to be 99->99.9%. Nevertheless, it is very possible to be able to obtain (-)-6 of 99->99.9% ee from 1 by a combination of the two processes, the photoreaction of 1 in an inclusion crystal with 3 and the acetalization of the resulted optically active ketone (+)-2.

Some other highly enantioselective [2 + 2] photocyclization reactions of enone derivatives in an inclusion complex with optically active host compound have also been reported for tropolone alkyl ether,<sup>8</sup> pyridone,<sup>9</sup> cycloocta-2,4,6-trien-1-one,<sup>10,11</sup> cycloocta-2,4-dien-1-one,<sup>10,11</sup> and coumarin.<sup>12</sup>

By the same procedure applied to 1, 2:1 inclusion crystals of 3 with 7 were prepared. Photoirradiation of a 2:1 inclusion crystal of N-benzyl-N-isopropyl-3-oxo-2-cyclohexanecarboxamide (7d) with 3b in a water suspension for 4 h gave optically pure (-)-N-benzyl-2-aza-3,3-dimethyl-1,5-dioxospiro[3.5]nonane (9d) as an oil in 53% yield.



When a 2:1 inclusion crystal of 7d with 3a is irradiated, optically pure (-)-9d was obtained in 51% yield (Table III). It has been known that the photoirradiation of 7d in MeCN for 1.5 h gives *rac-N*-isopropyl-2-aza-3-phenyl-1,5-dioxospiro[3.5]nonane (10) in 14% yield, although the stereochemistry on the benzyl carbon is not determined.<sup>4</sup> The reason why the isopropyl group but not the benzyl group of 7d reacts with the cyclohexenone group in the inclusion crystal with 3 is not clear; however, it is believed that the isopropyl group attacks the cyclohexenone plane from one side selectively in the inclusion crystal. Furthermore, the stereochemistry on the chiral carbon at position 3 of 9b and 9c was also controlled in the photoreactions. The irradiation of a 2:1 inclusion crystal

Table III. Photocyclization of 7 in a 2:1 InclusionCompound with 3 in a Water Suspension

			product						
						9			
host-guest compd			11				[α] <sub>D</sub>	optical	
host 3	guest 7	mp (°C)		yield (%)	mp (°C)		yield (%)	(deg) (c CHCl <sub>3</sub> )	purity (% ee)
8	a	165-167				9a	a		
b	8	193-195				9 <b>a</b>	а		
a	b	154 - 157	11b	27	173-175	9b	18	-20 (0.6)	>99.9
b	b	188-190	11b	30		9b	20		
a	с	123-124	11c	60	168-170	9c	41	-22 (1.3)	>99.9
b	с	158-160	11c	67		9c	46		
a	d	107-109				9d	51	-31 (1.3)	>99.9
b	d	159-163				9d	53	-33 (1.6)	>99.9
b	е	133-140				9e	46	0	
a	f	95-97				9f	a		
b	f	not clear				9f	a		

<sup>a</sup> No reaction occurred.

of 7c with 3b gave (-)-9c which upon treatment with hydroxylamine gave the optically pure (-)-oxime derivative (11c) of (-)-9c in 67% yield. Hydrolysis of 11c with 10%  $H_2SO_4$  gave optically pure (-)-9c (Table III). By the same method, 7b gave the optically pure (-)-oxime derivative (11b) of (-)-9c (Table III).

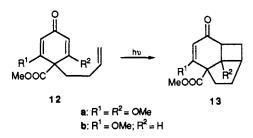
It is difficult to evaluate the efficiency of the enantioselectivity on the carbon-3 of 9 from the data shown in Table III because the possibility of purification of 11 by an optical resolution during its preparation and isolation cannot be neglected. In any way, optically pure 9b and 9c can be prepared by the combination of photoirradiation of 7b and 7c, respectively, in their inclusion crystals with 3 and purification of the reaction product via oxime derivatives. Nevertheless, 7a and 7f were inert to the photoirradiation in 2:1 inclusion crystals with 3, although 7a gives rac-9a by an irradiation in MeCN.<sup>4</sup> In the case of 7e, irradiation of a 2:1 inclusion crystal gave an optically inactive single product (9e) which consisted of two enantiomeric isomers on the spiro carbon, and the photoreaction of 7e in MeCN gives the same product in 70% yield.<sup>4</sup> On the reaction in MeCN, stereochemistry on the benzylcarbon of 7e is well controlled and only the stereoisomer which has the phenyl group on the azetidinone ring and the cyclohexanone carbonyl group in an anti-position is formed, although the stereochemistry on the spirocarbon is, of course, not controlled.<sup>4</sup>

In the inclusion crystal of 7a and 7f with 3, the methyl of 7a and the propyl group of 7f might be located at the position apart from the cyclohexenone moiety which abstracts hydrogen from these alkyl groups, and then 7a and 7f in the inclusion crystal might be inert to the photoreaction. In the inclusion crystal of 7e with 3b, the methyl group is located at a close position by the cyclohexenone moiety, and the diradical intermediate (8e) is formed by a hydrogen abstraction. However, the stereochemistry on the spiro-carbon of 9e was not controlled by the host 3b during the cyclization of 8e to 9e. Two reasons are considerable for the nonstereoselective photocyclization of 7e. One reason is that the generation of 8e from 7e in the inclusion crystal is followed by decomposition of the inclusion compound, and the cyclization of 8e to 9e occurs out of the inclusion crystalline lattice. Another reason is that the steric course of the cyclization of the initially formed 8e to 9e in the inclusion crystal with 3b is simply not affected by the chirality of the host 3b.

<sup>(8)</sup> Toda, F.; Tanaka, K. J. Chem. Soc., Chem. Commun. 1986, 1429.
(9) Tanaka, K.; Toda, F. J. Chem. Soc., Chem. Commun. 1983, 593.
(10) Toda, F.; Tanaka, K.; Oda, M. Tetrahedron Lett. 1988, 29, 653.
(11) Fujiwara, T.; Nanba, N.; Hamada, K.; Toda, F.; Tanaka, K. J.

Org. Chem. 1990, 55, 4532. (12) Tanaka, K.; Toda, F. J. Chem. Soc., Perkin Trans. 1 1992, 943.

Photocyclization reaction of 4-cabomethoxy-4-(3-butenyl)cyclohexa-2,5-dien-1-one derivatives such as 12 to rac-1-cabomethoxytricyclo[ $4.3.1.0^{7,10}$ ]dec-2-en-4-one (13) has been reported.<sup>5</sup> Enantioselective photocyclization of 12 to 13 was also achieved by irradiation of an inclusion crystal of 12 with 3 in the solid state. For example, reaction of



a powdered 2:1 inclusion crystal of 3,5-dimethoxy derivative 12a with 3a in a water suspension for 5 h gave (+)-13b of 73% ee in 50% yield. In the case of 3-methoxy derivative 12b, enantioselective inclusion complexation occurred to give a 1:1 complex of optically pure (-)-12b with 3a, and its irradiation in a water suspension gave optically pure (+)-13b in 57% yield. From the inclusion complex, optically pure (-)-12b was isolated. Although photoreaction in solution of the optically pure enantiomer of 12b, which had been prepared by a classical and complex resolution method, to the optically pure 13b has also been reported,<sup>5</sup> the inclusion complexation method is a much simpler and easier preparation method of the optically active 13b.

## **Experimental Section**

Preparation of inclusion compounds with the host **3a** and **3b** was carried out in the solvents ether and toluene-hexane (1:1), respectively, unless otherwise stated. Photoirradiations were carried out through a Pyrex filter by using a 100-W Hg lamp at room temperature (25 °C). IR spectra were measured with an IR spectrometer, Hitachi 260-10. All  $[\alpha]_D$  values were measured with a digital polarimeter, JASCO DIP-140. Optical purities were determined by HPLC using hexane-2-propanol (9:1) solvent unless otherwise stated (flow rate: 1.0 mL/min) and a column (0.46 cm  $\times$  25 cm) containing the chiral solid phase, Chiralcel OJ or OD and Chiralpak AS which are commercially available from Daicel Chemical Industries Ltd., Himeji, Japan.

**Preparation of 1.** The following compounds were prepared by a known literature procedure.<sup>3</sup> 2-(*N*-benzoyl-*N*-allylamino)cyclohex-2-enone (1a) (IR (neat)  $\nu_{max}$  1690 and 1660 cm<sup>-1</sup>); 2-(*N*-(*m*-chlorobenzoyl)-*N*-allylamino)cyclohex-2-enone (1b) (IR (neat)  $\nu_{max}$  1690 and 1655 cm<sup>-1</sup>); and 2-(*N*-(*p*-methylbenzoyl)-*N*-allylamino)cyclohex-2-enone (1c) (IR (neat)  $\nu_{max}$  1690 and 1650 cm<sup>-1</sup>). Since 1a-c were obtained as an oily material, they were identified by IR and <sup>1</sup>H NMR spectra and elemental analysis of their inclusion compounds with the host 3.

Preparation of Inclusion Compounds of 1 with 3. When a solution of 3a (1 g, 2 mmol) and 1a (0.27 g, 1 mmol) in ether (10 mL) was kept at room temperature for 2 h, a 2:1 inclusion compound of 3a and 1a was obtained as colorless needles (1.2 g, 93% yield, mp 118–119 °C): IR (Nujol)  $\nu_{max}$  3300, 1685, and 1615 cm<sup>-1</sup>. Anal. Calcd for C<sub>82</sub>H<sub>81</sub>NO<sub>10</sub>: C, 79.40; H, 6.58; N, 1.13. Found: C, 79.50; H, 6.63; N, 1.02. By the same procedure, the following inclusion compounds were prepared. A 2:1 inclusion compound of 3b with 1a (95% yield, mp 139.5–142 °C): IR (Nujol)  $\nu_{max}$  3300, 1690, and 1610 cm<sup>-1</sup>. Anal. Calcd for C<sub>84</sub>H<sub>84</sub>NO<sub>10</sub>: C, 79.53; H, 6.75; N, 1.10. Found: C, 79.42; H, 6.68; N, 1.05. A 2:1 inclusion compound of 3b with 1b (mp is not clear, 96% yield): IR (neat)  $\nu_{max}$  3330, 1690, and 1620 cm<sup>-1</sup>. Anal. Calcd for C<sub>84</sub>H<sub>84</sub>NO<sub>10</sub>CI: C, 77.43; H, 6.50; N, 1.07. Found: C, 77.56; H, 6.69; N, 1.07. A 2:1 inclusion compound of 3b with 1c (mp is not clear, 95% yield): IR (Nujol)  $\nu_{max}$  3350, 1690, and 1615 cm<sup>-1</sup>. Anal. Calcd for C<sub>86</sub>H<sub>87</sub>NO<sub>10</sub>: C, 79.60; H, 6.84; N, 1.09. Found: C, 79.32, H, 7.10; N, 1.03.

**Photocyclization of 1 to 2.** A suspension of a powdered 2:1 inclusion compound of 3a with 1a (1.2 g, 0.97 mmol) in water (120 mL) containing sodium alkylsulfate (0.1 g) as a surfactant was irradiated under stirring for 17 h. The reaction product was filtered, dried, and chromatographed on silica gel using AcOEthexane (1:1) as an eluent to give (+)-2a (0.16 g, 0.62 mmol, 64% yield, mp 106–107 °C): IR (Nujol)  $\nu_{\rm max}$  1720 and 1650 cm<sup>-1</sup>. Anal. Calcd for C<sub>16</sub>H<sub>17</sub>NO<sub>2</sub>: C, 75.27; H, 6.71; N, 5.49. Found: C, 75.25; H, 6.79; N, 5.28. By the same procedure, 2:1 inclusion compounds of 3b with 1a, 3b with 1b, and 3b with 1c gave (+)-2a, (+)-2b, and (+)-2c, respectively, in the yields shown in Table I. IR and analytical data of 2b and 2c are as follows. 2b: IR (Nujol)  $\nu_{max}$ 1725 and 1660 cm<sup>-1</sup>. Anal. Calcd for C<sub>16</sub>H<sub>16</sub>NO<sub>2</sub>Cl: C, 66.32; H, 5.57; N, 4.83. Found: C, 66.47; H, 5.60; N, 4.69. 2c: IR (Nujol)  $\nu_{\rm max}$  1720 and 1640 cm<sup>-1</sup>. Anal. Calcd for C<sub>17</sub>H<sub>19</sub>NO<sub>2</sub>: C, 75.81; H, 7.11; N, 5.20. Found: C, 76.01; H, 7.27; N, 5.01.

(-)-2,2-(Ethylenedioxy)-9-benzoyl-9-azatricyclo[5.2.1.0<sup>1,6</sup>]decane ((-)-6a) and Its Derivatives (-)-6b and (-)-6c. A solution of 2a (0.38 g, 1.49 mmol) and ethylene glycol (0.23 g, 3.71 mmol) in benzene (20 mL) containing a catalytic amount of TsOH was refluxed for 12 h under continuous removing of water by azeotropic distillation with benzene. The benzene solution was washed with water and dried over  $MgSO_4$ . The residue left after evaporation of the solvent was chromatographed on silica gel using AcOEt-hexane (1:1) as an eluent to give the acetal (-)-6a of 99% ee as colorless needles (0.32 g, 1.07 mmol, 72% yield, mp 108-110 °C): IR (Nujol) v<sub>max</sub> 1650 cm<sup>-1</sup>. Anal. Calcd for C<sub>18</sub>H<sub>21</sub>-NO3: C, 72.22; H, 7.07; N, 4.68. Found: C, 72.02; H, 7.22; N, 4.61. By the same procedure, the acetals (-)-6b of >99.9% ee and (-)-6c of 99% ee were obtained from the ketones (+)-2b and (+)-2c, respectively, in the yields shown in Table II. IR and analytical data are as follows. (-)-6b: IR (Nujol)  $\nu_{max}$  1670 and 1660 cm<sup>-1</sup>. Anal. Calcd for C<sub>18</sub>H<sub>20</sub>NO<sub>3</sub>Cl: C, 64.77; H, 6.04; N, 4.20. Found: C, 64.87, H, 6.14; N, 4.03. (-)-6c: IR (Nujol) v<sub>max</sub> 1660 and 1650 cm<sup>-1</sup>. Anal. Calcd for C<sub>19</sub>H<sub>23</sub>NO<sub>3</sub>: C, 72.82; H, 7.40; N, 4.47. Found: C, 73.02; H, 7.57; N, 4.26.

**Preparation of 7.** The following compounds were prepared by a modified condensation method of unsaturated oxoacid and amines using diethylphosphoryl cyanide (DEPC) instead of dicyclohexylcarbodiimide (DCC) in the reported method.4 N-(benzyl-N-methyl-3-oxo-3-cyclohexenecarboxamide (7a) (IR (neat)  $\nu_{max}$  1680 and 1640 cm<sup>-1</sup>); N-benzyl-N-ethyl-3-oxo-2cyclohexenecarboxamide (7b) (IR (neat)  $\nu_{max}$  1680 and 1640 cm<sup>-1</sup>); N-benzyl-N-n-propyl-3-oxo-2-cyclohexenecarboxamide (7c) (IR (neat)  $\nu_{max}$  1680 and 1640 cm<sup>-1</sup>); N-benzyl-N-isopropyl-3-oxo-2cyclohexenecarboxamide (7d) (IR (neat)  $\nu_{max}$  1680 and 1635 cm<sup>-1</sup>); N-dibenzyl-3-oxo-2-cyclohexenecarboxamide (7e) (IR (Nujol)  $\nu_{max}$ 1680 and 1640 cm<sup>-1</sup>); and N-benzyl-N-n-butyl-3-oxo-2-cyclohexenecarboxamide (7f) (IR (neat)  $\nu_{max}$  1685 and 1640 cm<sup>-1</sup>). Since 7a-7d,f were obtained as oily materials, they were identified by their IR spectra and elemental analysis of their inclusion compounds with the host 3.

Preparation of Inclusion Compounds of 7 with 3. When a solution of 3a (1.34 g, 2.71 mmol) and 7a (0.33 g, 1.36 mmol) in ether (50 mL) was kept at room temperature for 0.5 h, a 2:1 inclusion compound of 3a and 7a was obtained as colorless crystals (1.5 g, 90% yield, mp 165-167 °C): IR (Nujol) vmax 3300, 1665, and 1615 cm<sup>-1</sup>. Anal. Calcd for C<sub>81</sub>H<sub>81</sub>NO<sub>10</sub>: C, 79.19; H, 6.65; N, 1.14. Found: C, 79.01; H, 6.84; N, 1.26. By the same procedure, the following inclusion compounds were prepared. A 2:1 inclusion compound of **3b** with **7a** (92% yield, mp 193–195 °C): IR (Nujol)  $\nu_{max}$  3300, 1665, and 1615 cm<sup>-1</sup>. Anal. Calcd for C<sub>83</sub>H<sub>85</sub>NO<sub>10</sub>: C, 79.33; H, 6.82; N, 1.11. Found: C, 79.34; H, 7.06; N, 1.23. A 2:1 inclusion compound of 3a with 7b (91% yield, mp 154-157 °C): IR (Nujol)  $\nu_{max}$  3300, 1670, and 1610 cm<sup>-1</sup>. Anal. Calcd for C82H83NO10: C, 79.26; H, 6.73; N, 1.13. Found: C, 79.30; H, 6.99; N, 1.09. A 2:1 inclusion compound of 3b with 7b (91% yield, mp 188-190 °C): IR (Nujol) v<sub>max</sub> 3300, 1670, and 1610 cm<sup>-1</sup>. Anal. Calcd for  $C_{84}H_{67}NO_{10}$ : C, 79.40; H, 690; N, 1.10. Found: C, 79.46; H, 7.06: N, 1.17. A 2:1 inclusion compound of 3a with 7c (88% yield, mp 123–124 °C): IR (Nujol)  $\nu_{max}$  3300, 1670, and 1610 cm<sup>-1</sup>. Anal. Calcd for  $C_{88}H_{65}NO_{10}$ : C, 79.33; H, 6.82; N, 1.11. Found: C, 79.14; H, 6.96; N, 1.31. A 2:1 inclusion compound of **3b** with 7c (85% yield, mp 158–160 °C): IR (Nujol)  $\nu_{max}$  3300,

1670, and 1610 cm<sup>-1</sup>. Anal. Calcd for C<sub>85</sub>H<sub>89</sub>NO<sub>10</sub>: C, 79.47; H, 6.98; N, 1.09. Found: C, 79.32; H, 7.09; N, 1.26. A 2:1 inclusion compound of 3a with 7d (72% yield, mp 107-109 °C): IR (Nujol) νmax 3350, 1670, and 1610 cm<sup>-1</sup>. Anal. Calcd for C<sub>83</sub>H<sub>85</sub>NO<sub>10</sub>: C, 79.33; H. 6.82; N. 1.11. Found: C. 79.53; H. 6.70; N. 1.26. A 2:1 inclusion compound of 3b with 7d (80% yield, mp 159-163 °C): IR (Nujol)  $\nu_{max}$  3350, 1670, and 1610 cm<sup>-1</sup>. Anal. Calcd for C<sub>85</sub>H<sub>89</sub>NO<sub>10</sub>: C, 79.47; H, 6.98; N, 1.09. Found: C, 79.24; H, 7.28; N, 0.95. A 2:1 inclusion compound of 3b with 7e (75% yield, mp 133-140 °C): IR (Nujol) v<sub>max</sub> 3300, 1670, and 1620 cm<sup>-1</sup>. Anal. Calcd for C<sub>89</sub>H<sub>89</sub>NO<sub>10</sub>: C, 80.21; H, 6.73; N, 1.05. Found: C, 79.92; H, 6.96; N, 1.10. A 2:1 inclusion compound of 3a with 7f (50% yield, mp 95-97 °C): IR (Nujol) v<sub>max</sub> 3300, 1670, and 1610 cm<sup>-1</sup>. Anal. Calcd for C<sub>84</sub>H<sub>87</sub>NO<sub>10</sub>: C, 79.40; H, 6.90; N, 1.10. Found: C, 79.39; H, 7.06; N, 0.94. A 2:1 inclusion compound of 3b with 7f (62% yield, mp is not clear): IR (Nujol)  $\nu_{max}$  3300, 1670, and 1610 cm<sup>-1</sup>. Anal. Calcd for C<sub>86</sub>H<sub>91</sub>NO<sub>10</sub>: C, 79.54; H, 7.06; N, 1.08. Found: C, 79.68; H, 7.19; N, 0.77.

Photocyclization of 7 to 9. A suspension of a powdered 2:1 inclusion compound of 3b with 7b (6.72 g, 5.29 mmol) in water (120 mL) containing sodium alkylsulfate (0.1 g) as a surfactant was irradiated under stirring for 4 h. The reaction mixture was filtered, dried, and chromatographed on silica gel using AcOEthexane (1:2) as an eluent to give crude (-)-9b (1.18 g, 87% yield). Treatment of the crude (-)-9b (1.18 g) with NH<sub>2</sub>OH·HCl (0.32 g) and AcONa (0.66 g) in 50% aqueous EtOH (20 mL) gave the oxime 11b as colorless prisms (0.43 g, 34% yield, mp 173-175 °C): IR (Nujol)  $\nu_{max}$  3340 and 1730 cm<sup>-1</sup>. Anal. Calcd for  $C_{1e}H_{20}N_2O_2$ : C, 70.56; H, 7.40; N, 10.29. Found: C, 70.49; H, 7.58; N, 10.55. The overall yield of 11b from the 2:1 inclusion compound of 3b with 7b was 30%. After a mixture of 11b (0.43 g), 10% H<sub>2</sub>SO<sub>4</sub> (10 mL), and toluene (10 mL) was heated under reflux for 6 h, the toluene layer was separated, washed with water, and dried over MgSO<sub>4</sub>. The toluene solution was evaporated, and the residue was chromatographed on silica gel using AcOEthexane (1:2) as an eluent to give (-)-9b of >99.9% ee (0.27 g, 67%yield,  $[\alpha]_D - 20^\circ$  (c 0.6, CHCl<sub>3</sub>)): IR (neat)  $\nu_{max} 1740 \text{ cm}^{-1}$ . The overall yield of (-)-9b from the 2:1 inclusion compound of 3b with 7b was 20%. The optical purity of (-)-9b was determined by HPLC on the chiral solid phase, Chiralcel OJ. By the same method, (-)-11b was obtained from a 2:1 inclusion compound of 3a with 7b in 27% yield. Hydrolysis of the 11b gave (-)-9b of >99.9% ee in 18% overall yield from the inclusion compound. By the same procedure, a 2:1 inclusion compound of 3b with 7c gave 11c as colorless needles (67% yield from the inclusion compound, mp 168-170 °C): IR (Nujol) vmax 3300 and 1735 cm<sup>-1</sup>. Anal. Calcd for C17H22N2O2: C, 71.30; H, 7.95; N, 9.78. Found: C, 71.53; H, 7.95; N, 9.60. Hydrolysis of 11c gave (-)-9c of >99.9% ee (46% yield from the inclusion compound,  $[\alpha]_D -22^\circ$  (c 1.3, CHCl<sub>3</sub>)). From the 2:1 inclusion compound of 3a with 7c were obtained 11c and 9c in 60% and 41% overall yields, respectively. Similar photoirradiation of a 2:1 inclusion compound of 3b with 7d followed by similar workup as above gave (-)-7d of >99.9% ee as an oil (53% yield,  $[\alpha]_D$ -33° (c 1.6, CHCl<sub>3</sub>)): IR (neat)  $\nu_{max}$ 1730 cm<sup>-1</sup>. Anal. Calcd for C<sub>17</sub>H<sub>21</sub>NO<sub>2</sub>: C, 75.25; H, 7.80; N, 5.16. Found: C, 75.46; H, 7.99; N, 5.03. From a 2:1 inclusion compound of 3a with 7d, (-)-9d of >99.9% ee was obtained in 51% yield  $[\alpha]_D$  -31° (c 1.3, CHCl<sub>3</sub>)). The optical purity of 9c and 9d was also determined by HPLC on the chiral solid phase, Chiralcel OJ. By the same procedure, a 2:1 inclusion compound of 3b with 7e gave optically inactive 9e in 46% yield as colorless needles (mp 155-159 °C): IR (Nujol) v<sub>max</sub> 1750 and 1720 cm<sup>-1</sup>.

Anal. Calcd for  $C_{21}H_{21}NO_2$ : C, 78.97; H, 6.63; N, 4.39. Found: C, 79.02; H, 6.75; N, 4.18. However, 2:1 inclusion compounds of **3a** or **3b** with **7a** and **7f** are inert to the photoirradiation, and inclusion compounds were recovered unchanged.

Preparation of Inclusion Compounds of 12 with 3 and Optical Resolution of 12b. When a solution of 3a (0.47 g, 0.95 mmol) and 12a (0.13 g, 0.49 mmol) in ether-hexane (1:1, 10 mL) was kept at room temperature for 10 h, a 2:1 inclusion compound of 3a with 12a was obtained as colorless needles (0.36, g, 60%)yield, mp not clear): IR (Nujol)  $\nu_{max}$  3240, 1730, 1655, 1610, and 1600 cm<sup>-1</sup>. Anal. Calcd for C<sub>80</sub>H<sub>82</sub>O<sub>13</sub>: C, 76.78; H, 6.60. Found: C, 77.00; H, 6.59. When a solution of 3a (1.4 g, 2.9 mmol) and 12b (1.4 g, 5.9 mmol) in ether-hexane (1:1, 50 mL) was kept at room temperature for 10 h, a 1:1 inclusion compound of 3a with (-)-12b was obtained as colorless prisms (1.4g, 74% yield). Optical purity of the (-)-12b in the inclusion compound was found to be 66.4% ee by HPLC analysis on the chiral solid-phase Chiralcel OD. Three recrystallizations of the inclusion compound from ether-hexane (1:1) gave pure inclusion compound (0.95 g, 46%yield, mp 91-95 °C): IR (Nujol) v<sub>max</sub> 3270, 1760, 1735, 1655, and 1595 cm<sup>-1</sup>. Anal. Calcd for C46H47O8: C, 75.91; H, 6.51. Found: C, 75.88; H, 6.62. Chromatography of the pure inclusion compound on silica gel using AcOEt-toluene (1:9) as an eluent gave (-)-12b of >99.9% ee in quantitative yield ( $[\alpha]_D$  -76.0° (c 0.325, CHCl<sub>3</sub>) (lit.<sup>5</sup> [ $\alpha$ ]<sub>D</sub> -75.3° (c 0.17, CHCl<sub>3</sub>)).

However, neither 12a nor 12b formed an inclusion compound with the host 3b.

Photocyclization of 12 to 13. A suspension of powdered 2:1 inclusion compound of 3a with 12a (0.36 g) in water (100 mL) containing sodium alkylsulfate (0.1 g) as a surfactant was irradiated under stirring for 5 h. The reaction mixture was filtered, dried, and chromatographed on silica gel using AcOEthexane (1:1) as an eluent gave to give (+)-13a of 73% ee (0.04 g, 50% yield, [a]<sub>D</sub> +5.7°, mp 110-113 °C (lit.<sup>5</sup> mp for rac-13a 140-141 °C)). The structure of 13a was determined by comparison of its IR and <sup>1</sup>H NMR spectra with those reported.<sup>5</sup> The optical purity of the (+)-13a was determined by HPLC on the chiral solid phase, Chiralcel OJ. By the same photoirradiation procedure, a 1:1 inclusion compound of 3a with 12b of >99.9% ee gave (+)-13b of >99.9% ee in 57% yield ( $[\alpha]_D$  +45° (c 0.70, MeOH),  $+36.4^{\circ}$  (c 0.686, CHCl<sub>3</sub>) (lit.<sup>5</sup> [ $\alpha$ ]<sub>D</sub> +30.2° (c, 0.625, CHCl<sub>3</sub>), mp 110-111 °C (lit.<sup>5</sup> mp for rac-13b 76-78 °C)). The optical purity of (+)-13b was determined by HPLC on the chiral solid phase, Chiralpak As, using hexane-EtOH (95:5) as solvent.

Acknowledgment. We thank the Ministry of Education, Science and Culture, Japan, for Grant-in-Aid for Scientific Research (B), No, 04453102.

Supplementary Material Available: IR spectral data for inclusion compounds of 3a and 3b with 7a-f shown in Table III, inclusion compounds of 3a with 12a and 12b, and photocyclization products 9b-e and their oxime derivatives 11b and 11c and <sup>1</sup>H NMR spectral data measured with a JEOL JNM-PMX 60 spectrometer in CDCl<sub>3</sub> for inclusion compounds of 3a and 3b with 7a-f shown in Table III, photocyclization products 9b-e and their oxime derivatives 11b and 11c, and inclusion compounds of 3a with 12a and 12b (38 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.